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Key indicators

Single-crystal X-ray study T = 292 KMean $\sigma(\text{C}-\text{C}) = 0.004 \text{ Å}$ R factor = 0.068 wR factor = 0.174 Data-to-parameter ratio = 15.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

N-[4-(2,2-Dimethylpropionylamino)phenyl]-2,2-dimethylpropionamide

In the title compound, $C_{16}H_{24}N_2O_8$, the molecule is centrosymmetric. The crystal packing is stabilized by intermolecular $N-H\cdots O$ hydrogen bonds.

Comment

The luminescence features of some amide compounds have been studied by Frederick & Liu (1999). In this paper, we report the crystal structure of the title compound, (I), which is a new amide compound.



In the centrosymmetric molecule of (I) (Fig. 1), the bond lengths and angles present no unusual features. The crystal packing (Fig. 2) is stabilized by intermolecular hydrogen bonds (Table 1). The N···O distance of 2.14 (2) Å is shorter than those in benzanilide [2.34 (3) Å; Bowes *et al.*, 2003] and *N*-[4-(phenylethynyl)phenyl]benzamide [3.42 (2) Å; Yin *et al.*, 2005].

Experimental

To a mixture of benzene-1,4-diamine (0.54 g, 5 mmol) and Et_3N (5 ml) in tetrahydrofuran (30 ml), 2,2-dimethylpropionyl chloride (1.2 g, 10 mmol) was added dropwise and the mixture stirred at room temperature for 3 h. The solvent was then evaporated on a rotary evaporator and the mixture was extracted with EtOAc (50 ml) to obtain the title compound. Single crystals of (I) suitable for X-ray diffraction were obtained by slow evaporation of a solution in CH_2Cl_2 –MeOH (1:1 ν/ν).



Figure 1

A view of the molecule of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 50% probability level and H atoms are represented by circles of arbitrary size. [Symmetry code: (a) 2 - x, 1 - y, 2 - z]

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Crystal data

 $\begin{array}{l} C_{16}H_{24}N_2O_2\\ M_r = 276.37\\ \text{Monoclinic, } P2_1/c\\ a = 12.166 \ (4) \ \text{\AA}\\ b = 7.677 \ (3) \ \text{\AA}\\ c = 9.243 \ (3) \ \text{\AA}\\ \beta = 96.613 \ (6)^\circ\\ V = 857.6 \ (5) \ \text{\AA}^3\\ Z = 2 \end{array}$

Data collection

Bruker SMART 1K CCD areadetector diffractometer φ and ω scans Absorption correction: none 3943 measured reflections 1493 independent reflections

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0699P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.068$	+ 0.4858P]
$wR(F^2) = 0.174$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.09	$(\Delta/\sigma)_{\rm max} < 0.001$
1493 reflections	$\Delta \rho_{\rm max} = 0.25 \text{ e } \text{\AA}^{-3}$
94 parameters	$\Delta \rho_{\rm min} = -0.15 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	

 $D_x = 1.070 \text{ Mg m}^{-3}$

Cell parameters from 1225

Mo $K\alpha$ radiation

reflections

 $\theta = 3.1-23.6^{\circ}$ $\mu = 0.07 \text{ mm}^{-1}$

T = 292 (2) K

 $R_{\rm int} = 0.026$

 $\theta_{\rm max} = 25.0^{\circ}$ $h = -14 \rightarrow 12$

 $k = -9 \rightarrow 9$

 $l = -10 \rightarrow 9$

Block, colourless

 $0.30 \times 0.20 \times 0.10 \ \text{mm}$

1136 reflections with $I > 2\sigma(I)$

Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$N1-H1A\cdotsO1^{i}$	0.86	2.14	2.929 (3)	153
6	. 1 1			

Symmetry codes: (i) $x, -y + \frac{1}{2}, z - \frac{1}{2}$.

After their location in a difference Fourier map, H atoms were placed in calculated positions and allowed to ride on their parent atoms, with with C-H = 0.93 Å for aromatic C and 0.96 Å for methyl C, and N-H = 0.86 Å, and with $U_{iso}(H) = 1.2 U_{eq}(C)$ for aromatic C and N, or $1.5_{eq}(C)$ for methyl C.

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Sheldrick, 2001); software used to prepare material for publication: *SHELXTL*.

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Figure 2

 $N-H\cdots O$ intermolecular interactions (dashed lines) in the crystal structure of (I). H atoms have been omitted.

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